



**A new concept to assess the
environmental fate of chiral pesticides:
Combined stable isotope and
enantiomeric fractionation of alpha HCH**

Silviu-Laurentiu Badea

PhD Student

Department of Chemistry

Silviu.badea@chem.umu.se



Content

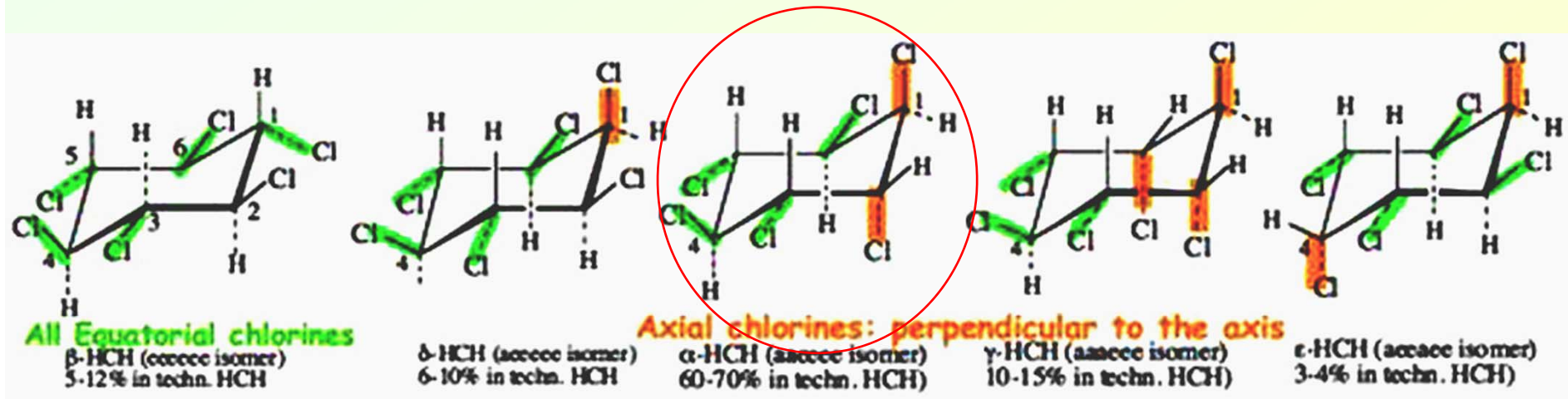
- **HCHs: A global problem**
- **Stable isotope fractionation: a tool for the assessment of the environmental fate of HCHs**
- **Chirality of alpha HCH**
- **Stable Isotope Analysis of alpha HCH's enantiomers**
- **Presentation of the Experimental Approach**
- **Results**
- **Conclusions**

HCHs: A GLOBAL PROBLEM



HCHs: A GLOBAL PROBLEM

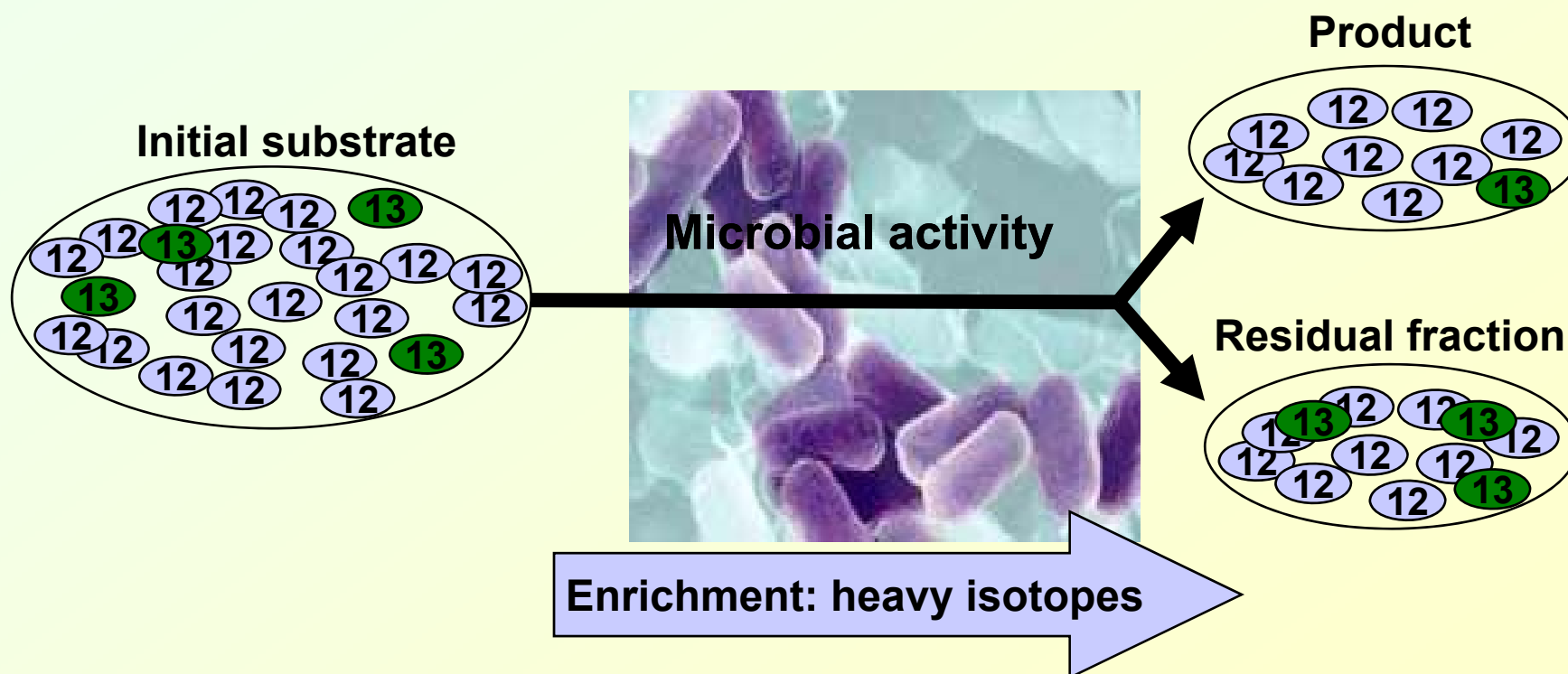
In the contaminated sites technical grade HCH contains mainly 5 isomers: α , β , γ , δ , ϵ



Alpha HCH is the only optical active isomer

Stable isotope fractionation: a tool for the assessment of the environmental fate of HCHs

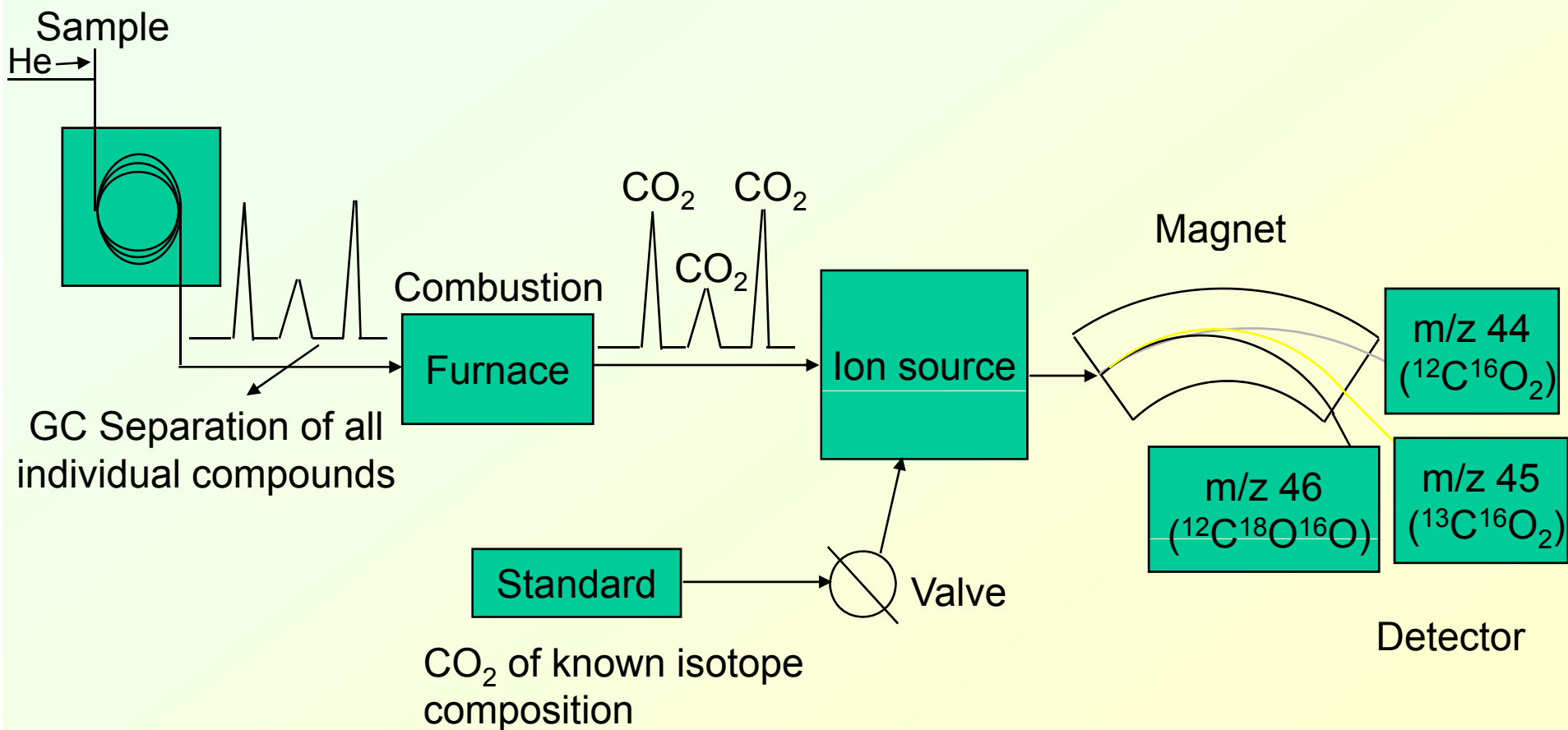
- **Compound Specific Isotope analysis (CSIA):** identification of biodegradation
- **Preferential use of the lighter isotope during biodegradation**



Compound-specific Stable Isotope Analysis (CSIA) of HCHs

Gas Chromatography (GC)

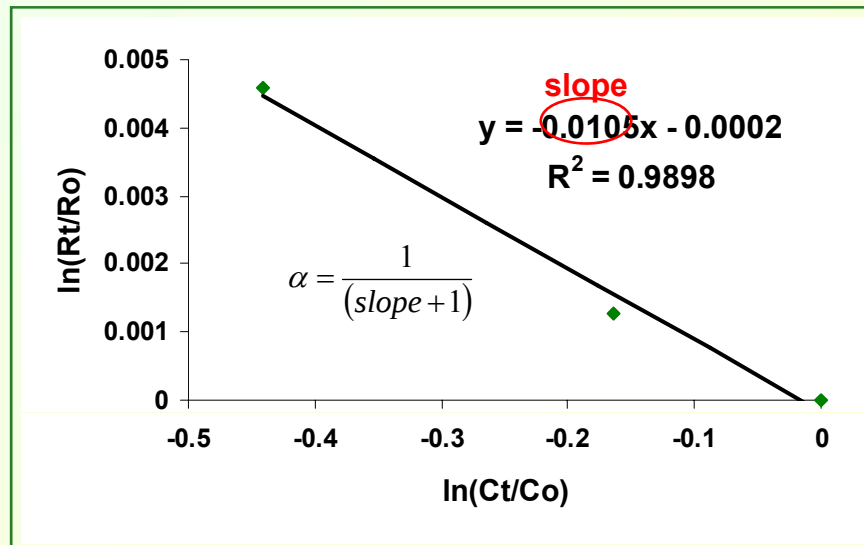
Isotope Ratio Mass Spectrometry (IRMS)



Quantification of stable isotope fractionation

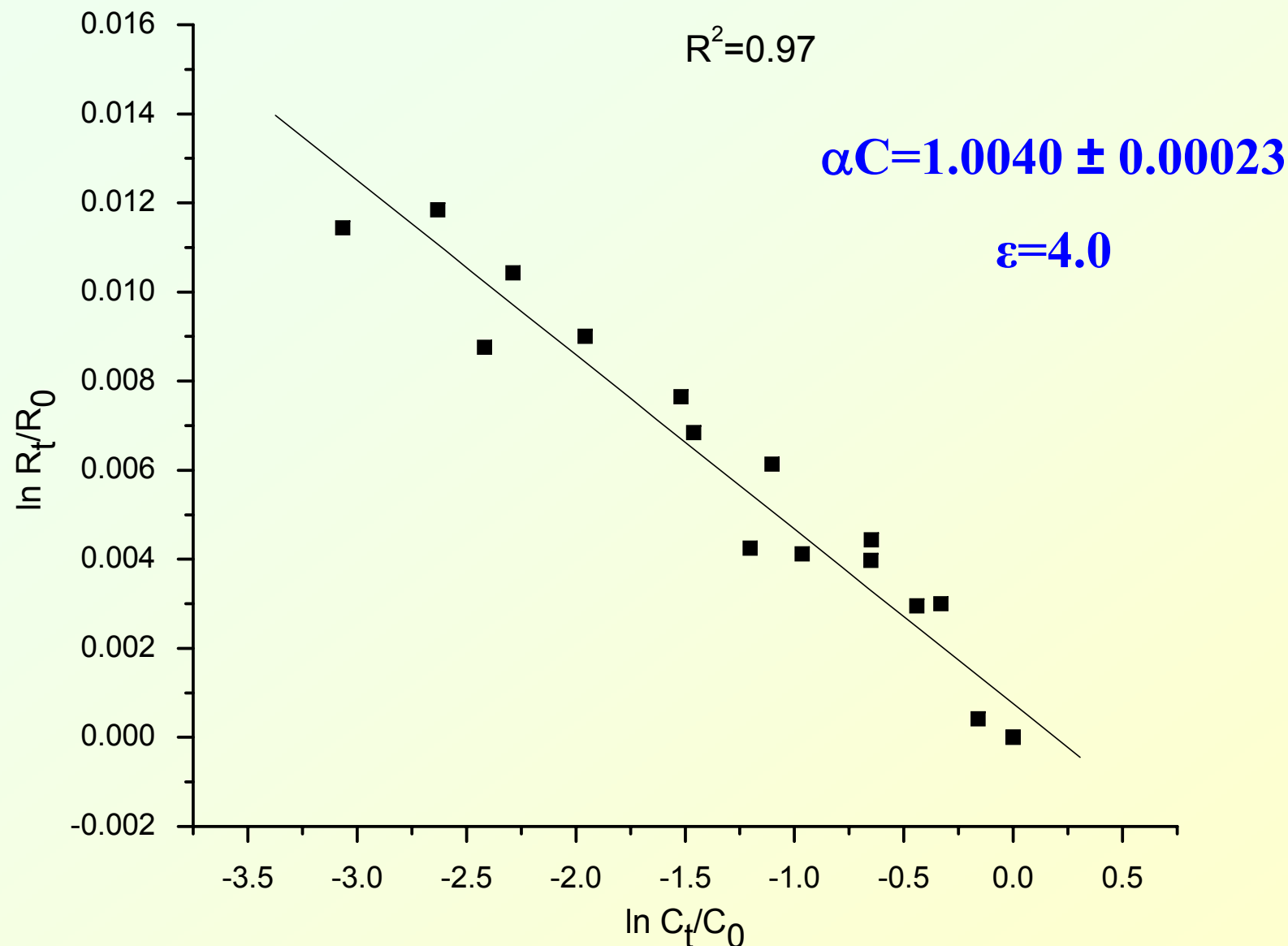
Rayleigh Equation:

$$\frac{R_t}{R_0} = \left(\frac{C_t}{C_0} \right)^{\left(\frac{1}{\alpha} - 1 \right)} \longrightarrow \ln \left(\frac{R_t}{R_0} \right) = \left(\frac{1}{\alpha} - 1 \right) * \ln \left(\frac{C_t}{C_0} \right)$$

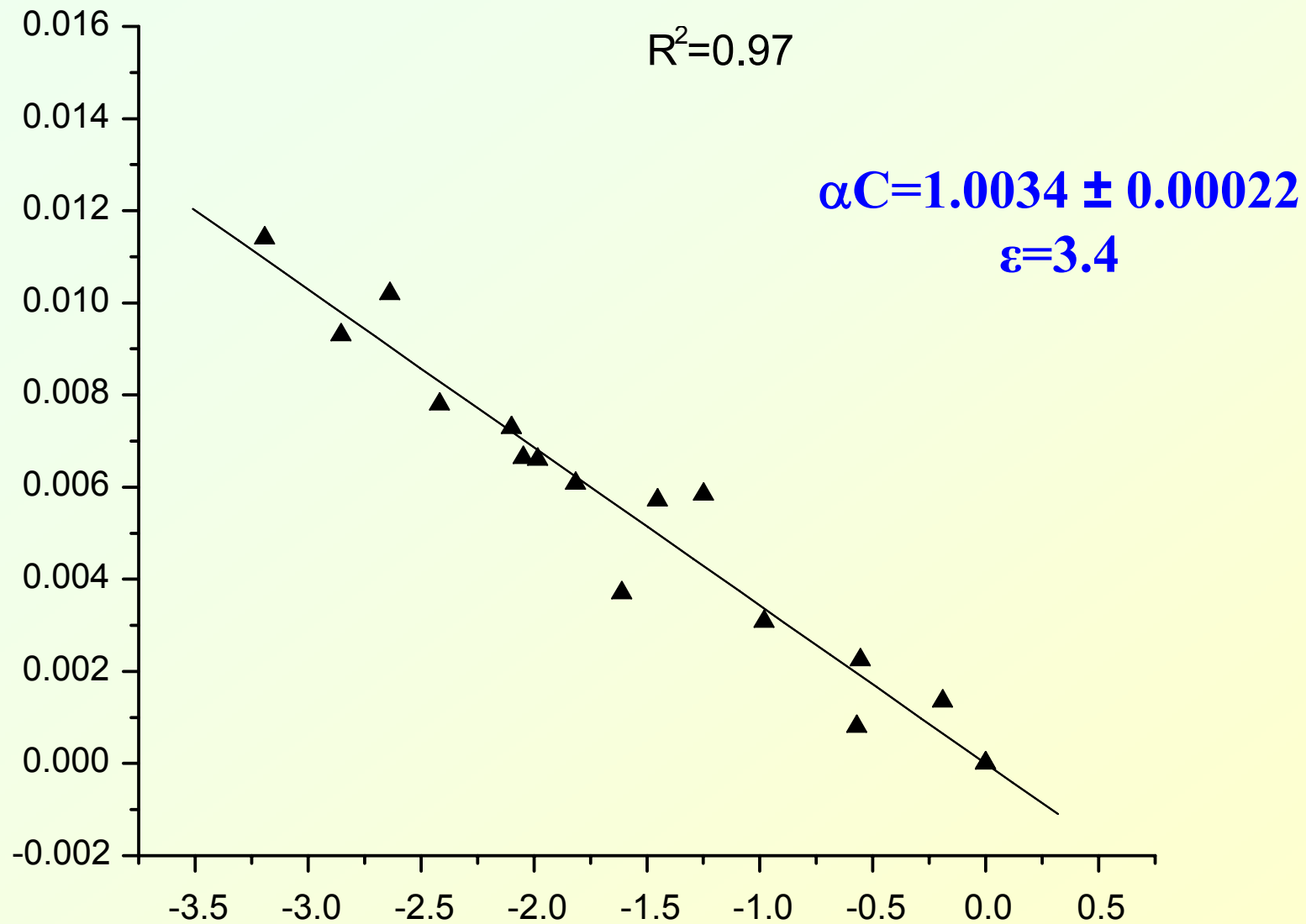


C_0 = initial concentration
 C_t = concentration at time t
 R_0 = initial isotope ratio
 R_t = isotope ratio at time t
 α = isotope fractionation factor

$$\varepsilon [‰] = (\alpha - 1) * 1000$$



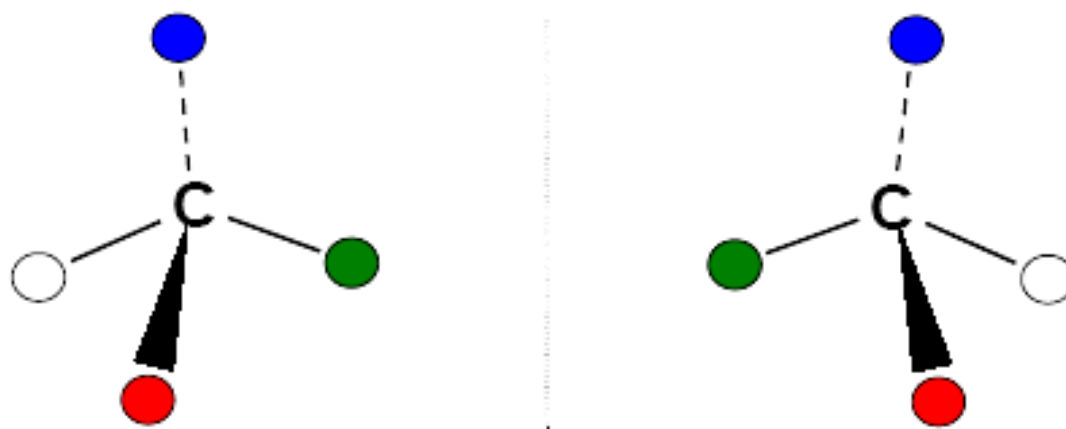
Quantitative assessment of isotope fractionation of γ HCH in the biodegradation experiments with *D. Gigas* (Badea et. al (2009), Environmental Science & Technology)



Quantitative assessment of isotope fractionation of γ HCH in the biodegradation experiments with *D. Multivorans* (Badea et. al (2009), Environmental Science & Technology)

Chirality of alpha HCH

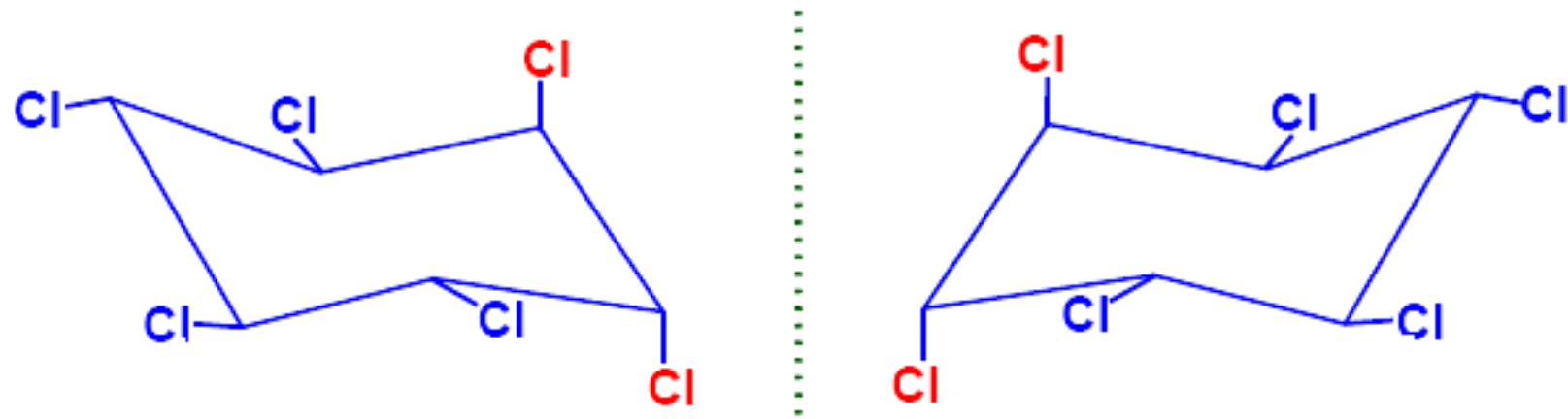
A chiral (handed) molecule cannot be superimposed on its mirror image



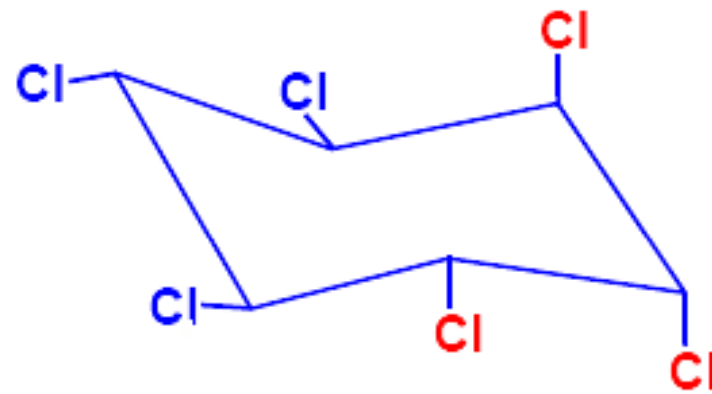
Types of chiral compounds:

1. The simplest chiral molecule has one asymmetric tetrahedral carbon atom (four different groups attached)

Types of chiral molecules
2. ring structures with no symmetry plane

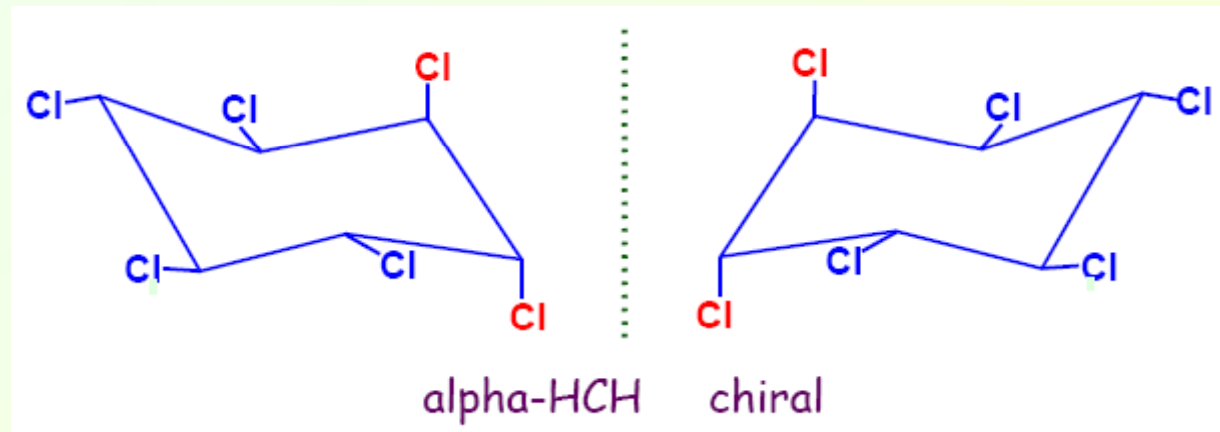


alpha-HCH chiral



gamma-HCH (lindane) achiral

Chirality of alpha HCH

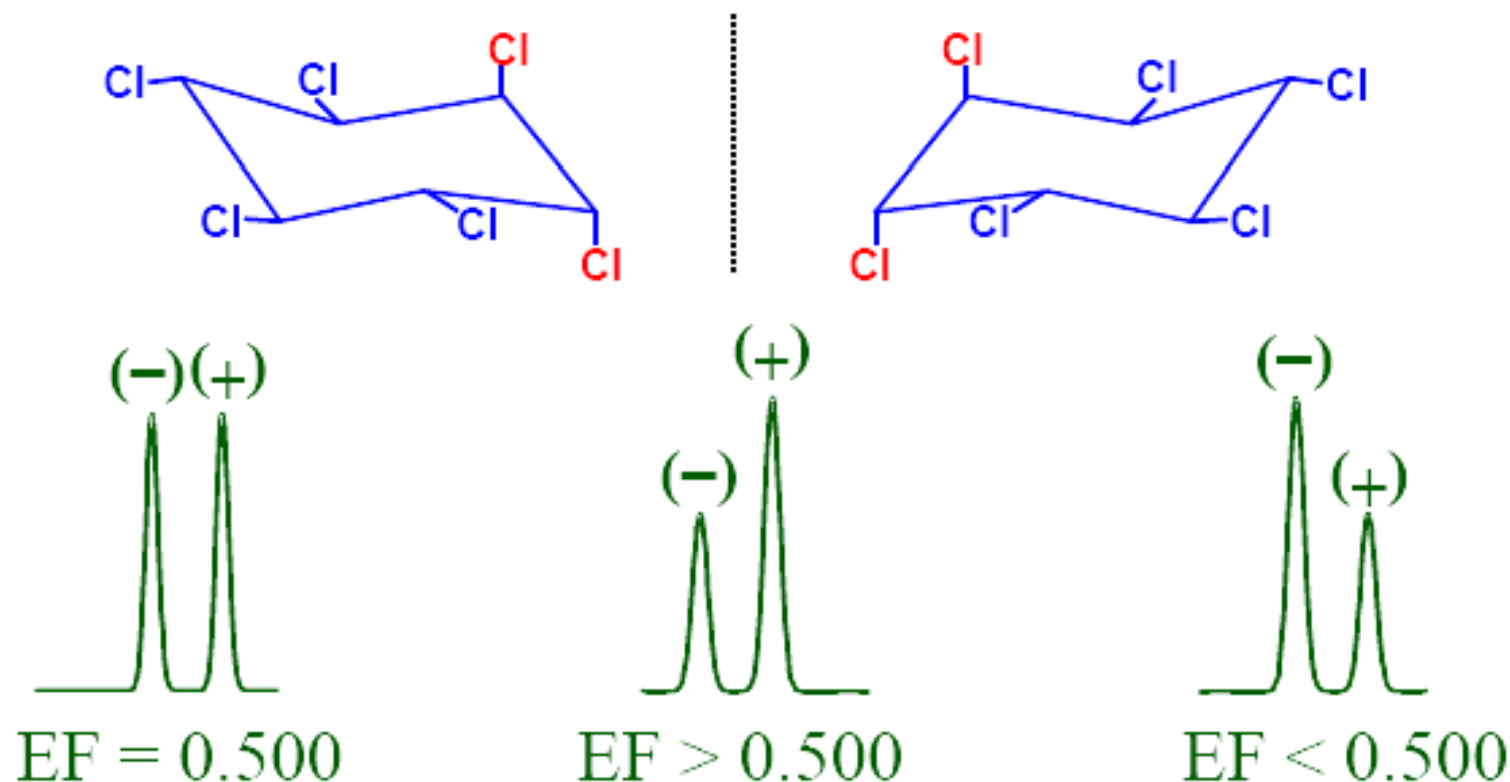


- Biodegradation: one enantiomer is usually preferentially degraded over the other

- Enantiomeric fraction (EF) of α -HCH

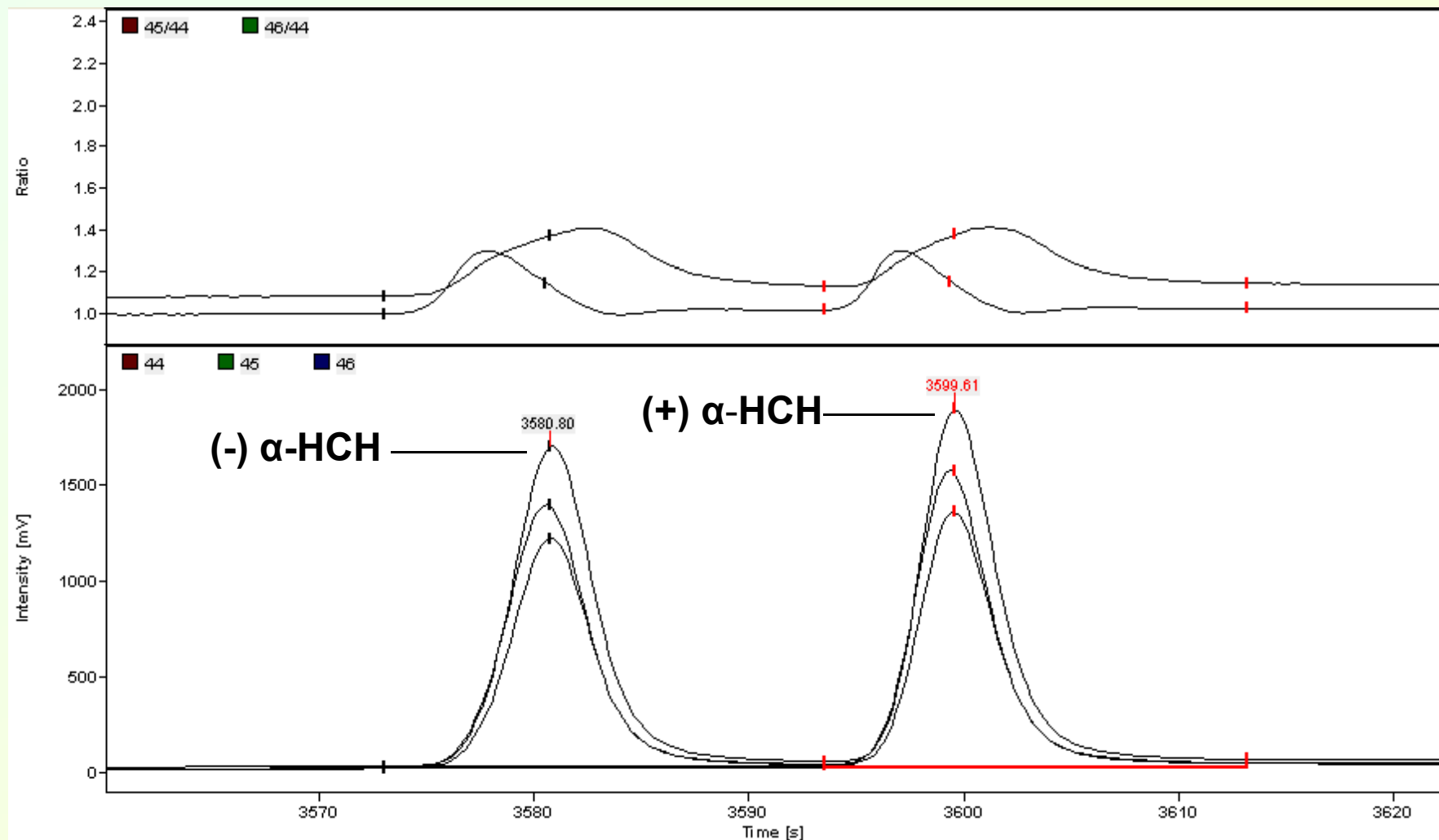
$$EF = (+) \alpha\text{-HCH} / [(+) \alpha\text{-HCH} + (-) \alpha\text{-HCH}]$$

α -HCH is chiral

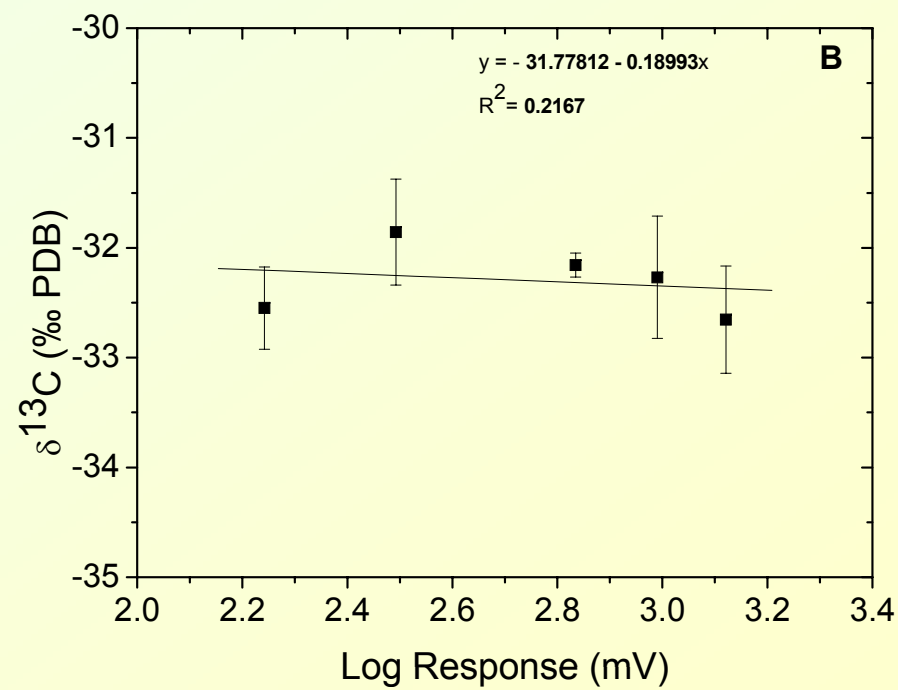
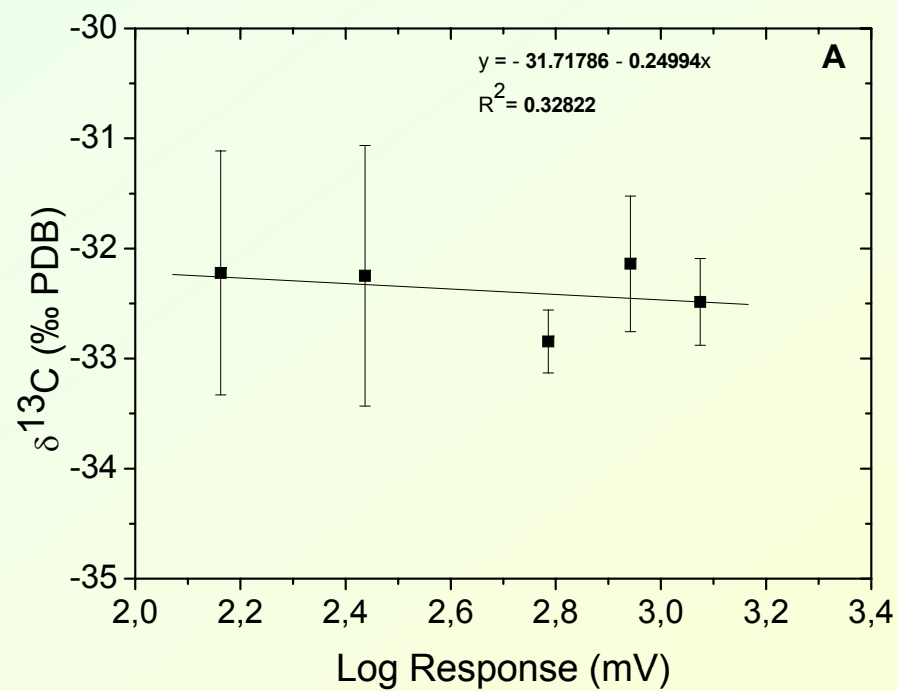


- Manufactured as a racemate
- Enantiomer fraction, $EF = (+)/[(+) + (-)] = 0.500$ (racemic)
- Preferential degradation of one enantiomer in water and soil leads to non-racemic composition

Stable Isotope Analysis of alpha HCH's enantiomers



The alpha HCH Enantiomers in a GC-C-IRMS Chiral Chromatogram



**The plot graphs of the linearity test for (-) α HCH (A) and (+) α HCH (B).
 The concentrations of the enantiomers pair were : 25, 50, 100, 150 and 200 μM**

Experimental Approach

■ Combined stable isotope and enantiomeric fractionation of alpha HCH with reference experiments with cultures obtained from culture collections (DSM) (*Clostridium pasteurianum* DSM 525, *Desulfococcus multivorans* DSM 2059

Desulfovibrio gigas DSM 1382) to obtain isotope fractionation factors for future field investigation

■ Alpha HCH's enantiomers quantification (GC-MS Chiral Analysis with γ -DEX™ Chiral Column)

■ Measurement of the isotope fractionation during degradation of HCH's enantiomers (GC-C-IRMS with β or γ -DEX™ Chiral Column)

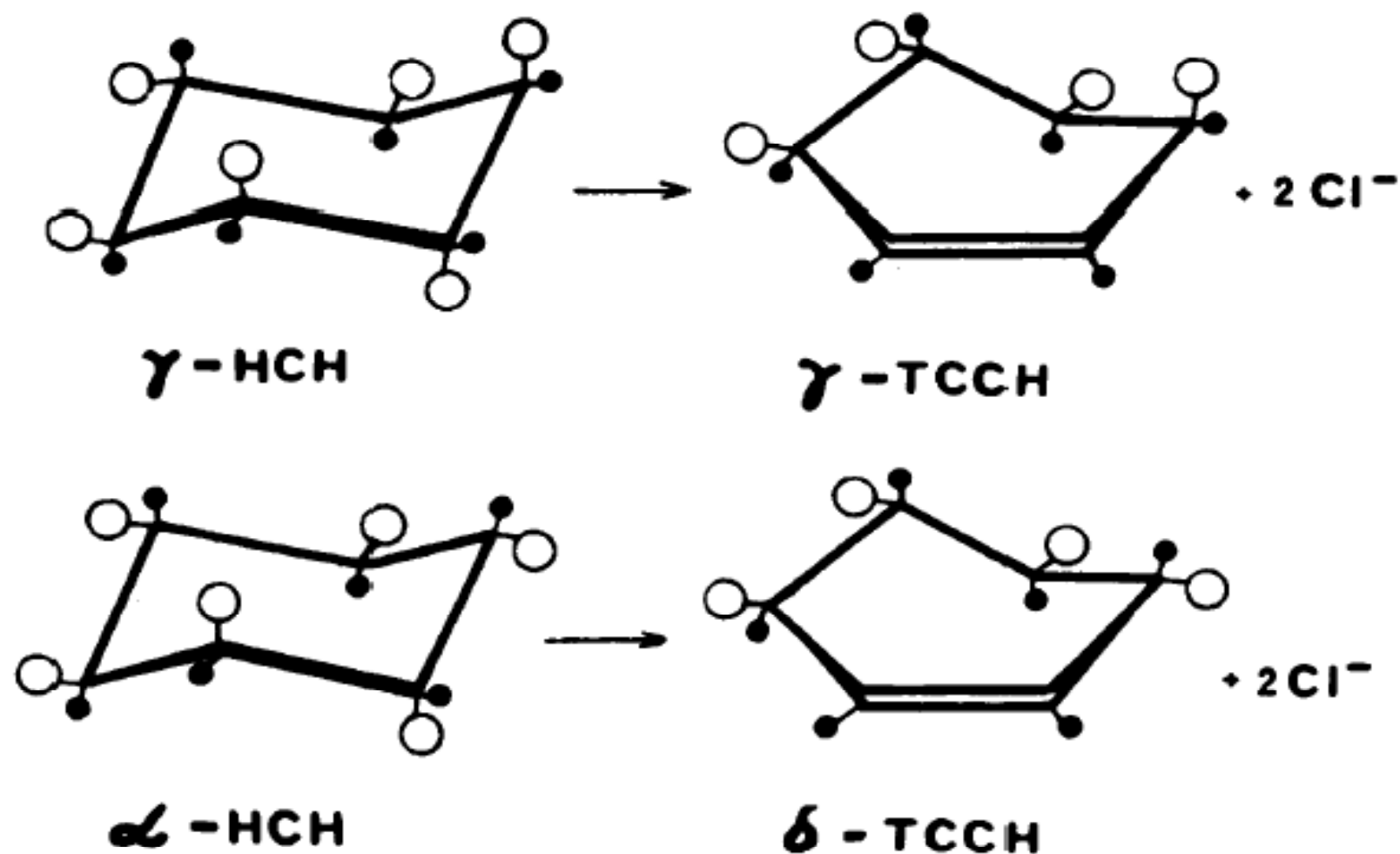
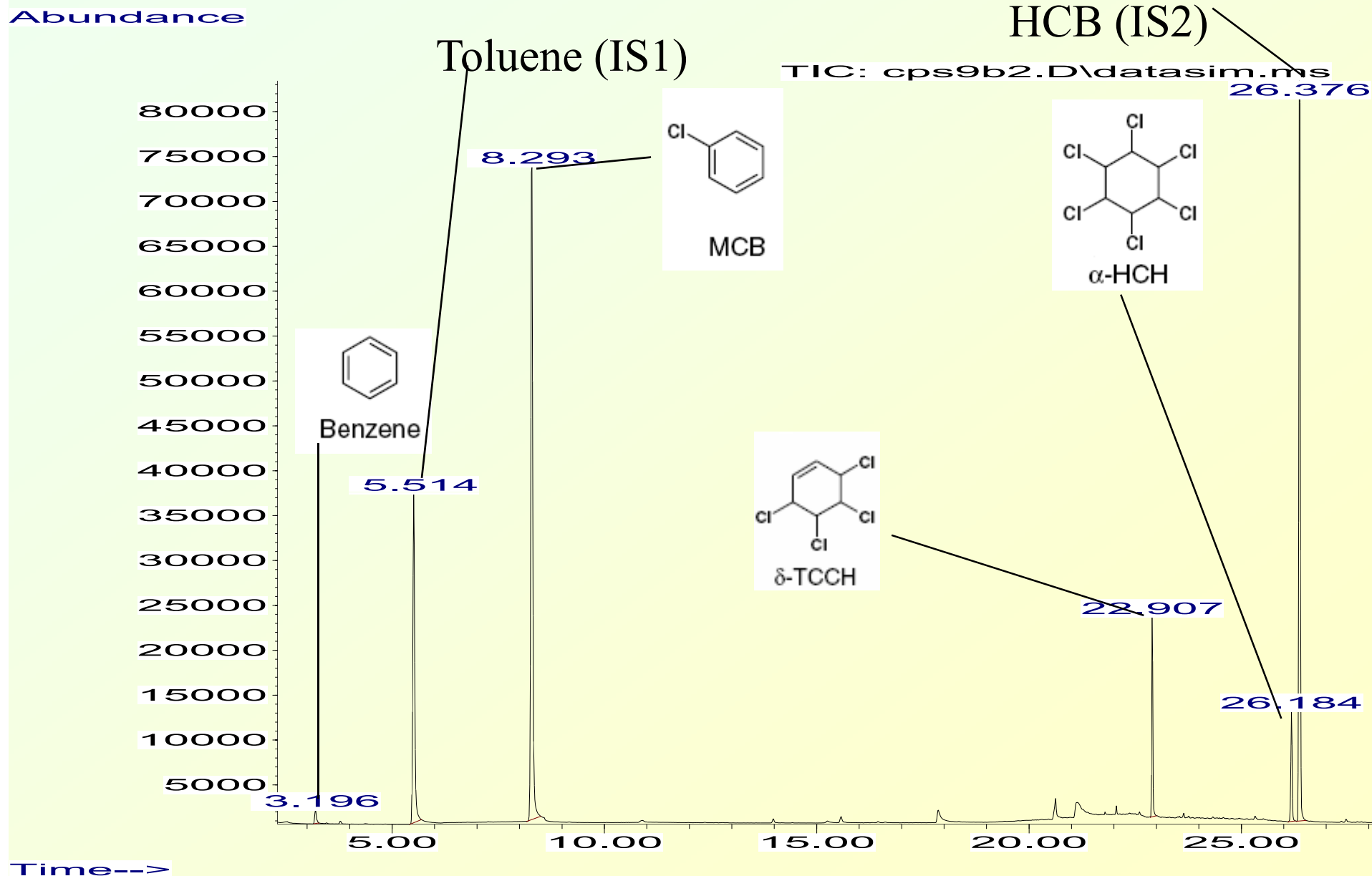
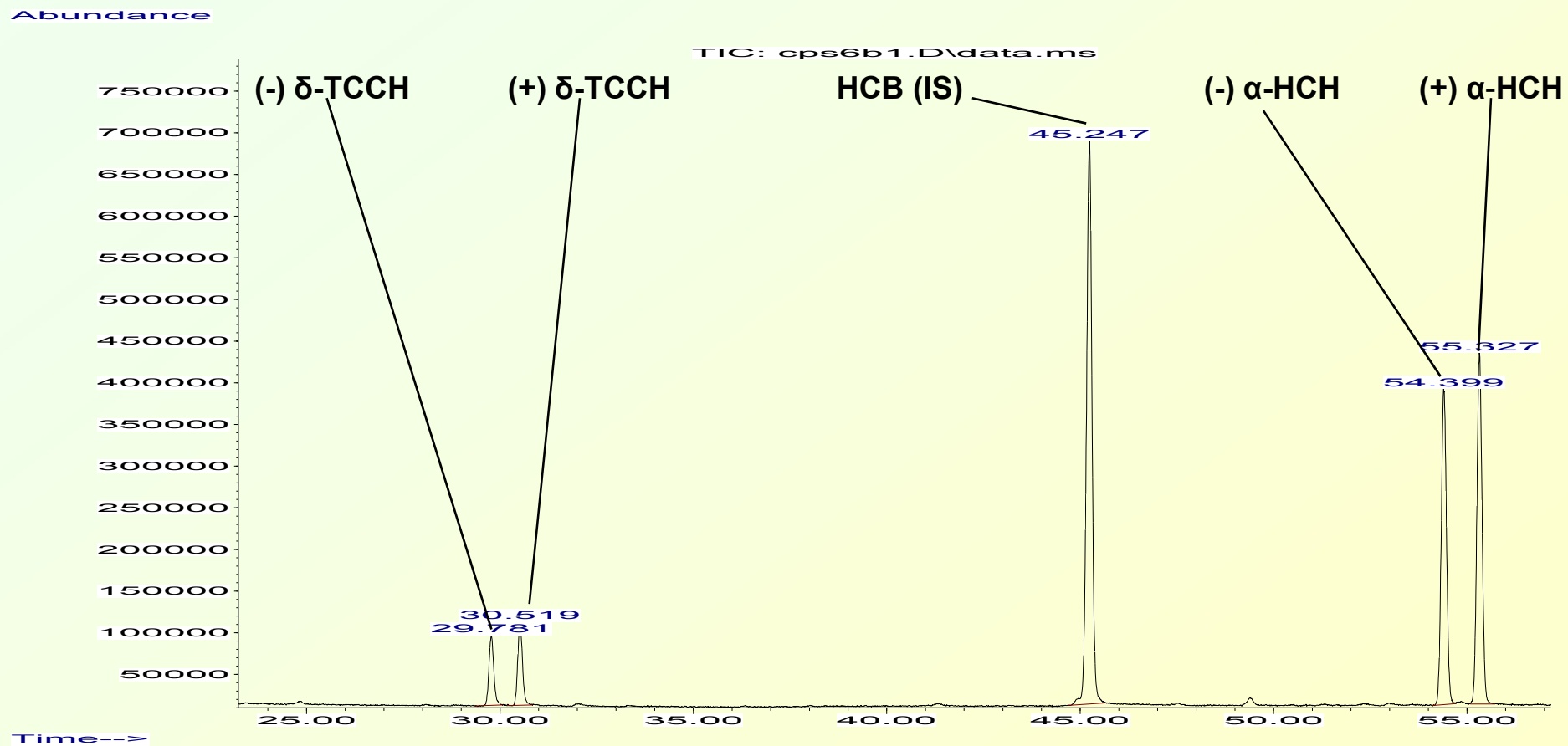


FIG. 1. Degradation of the α - and γ -isomers of HCH to the δ - and γ -isomers of TCCH by *C. sphenoides*. Symbols: Chlorine (○); hydrogen (●).

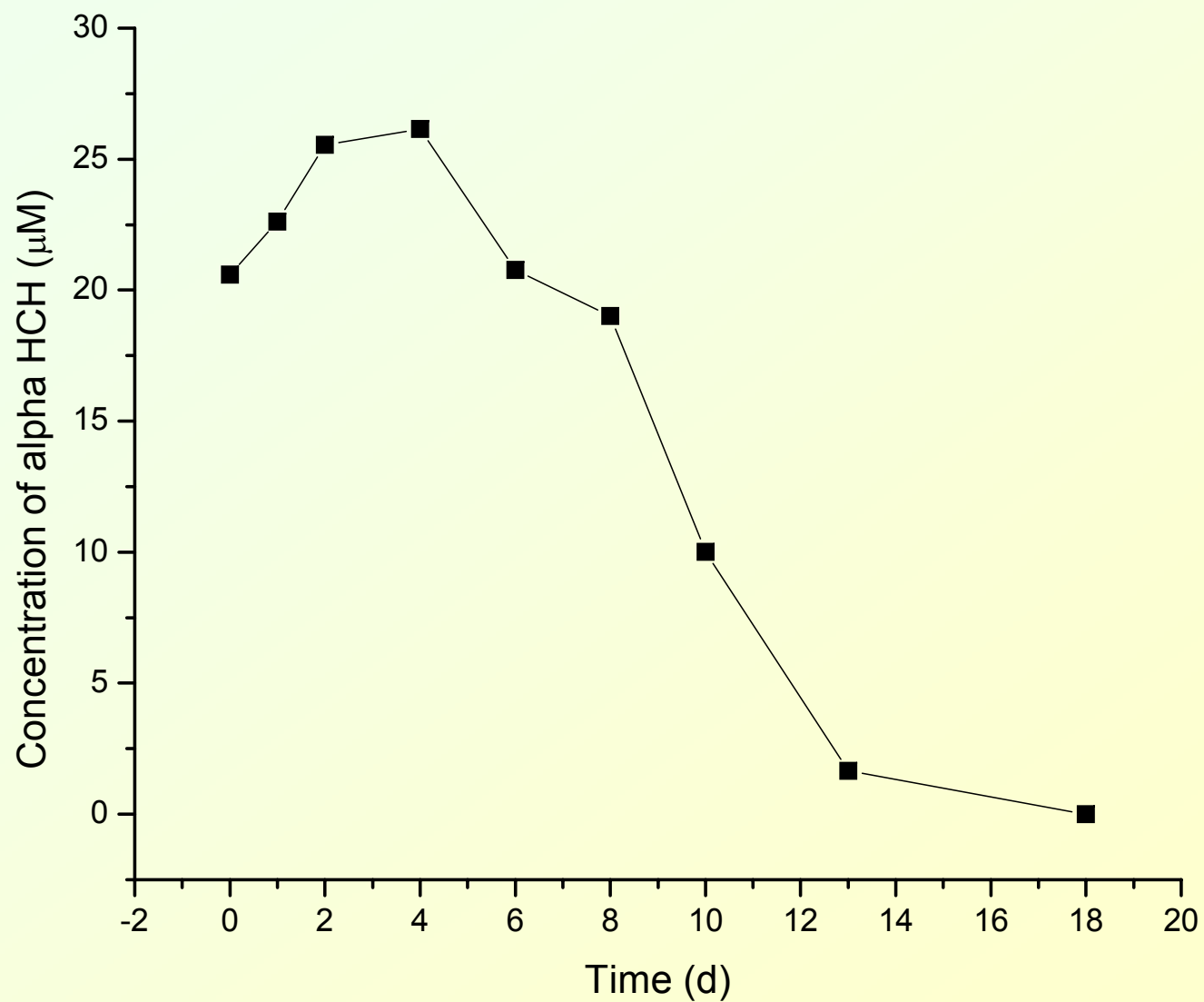
Detected metabolites in the Biodegradation experiments with *Clostridium Pasteurianum* DSM 525

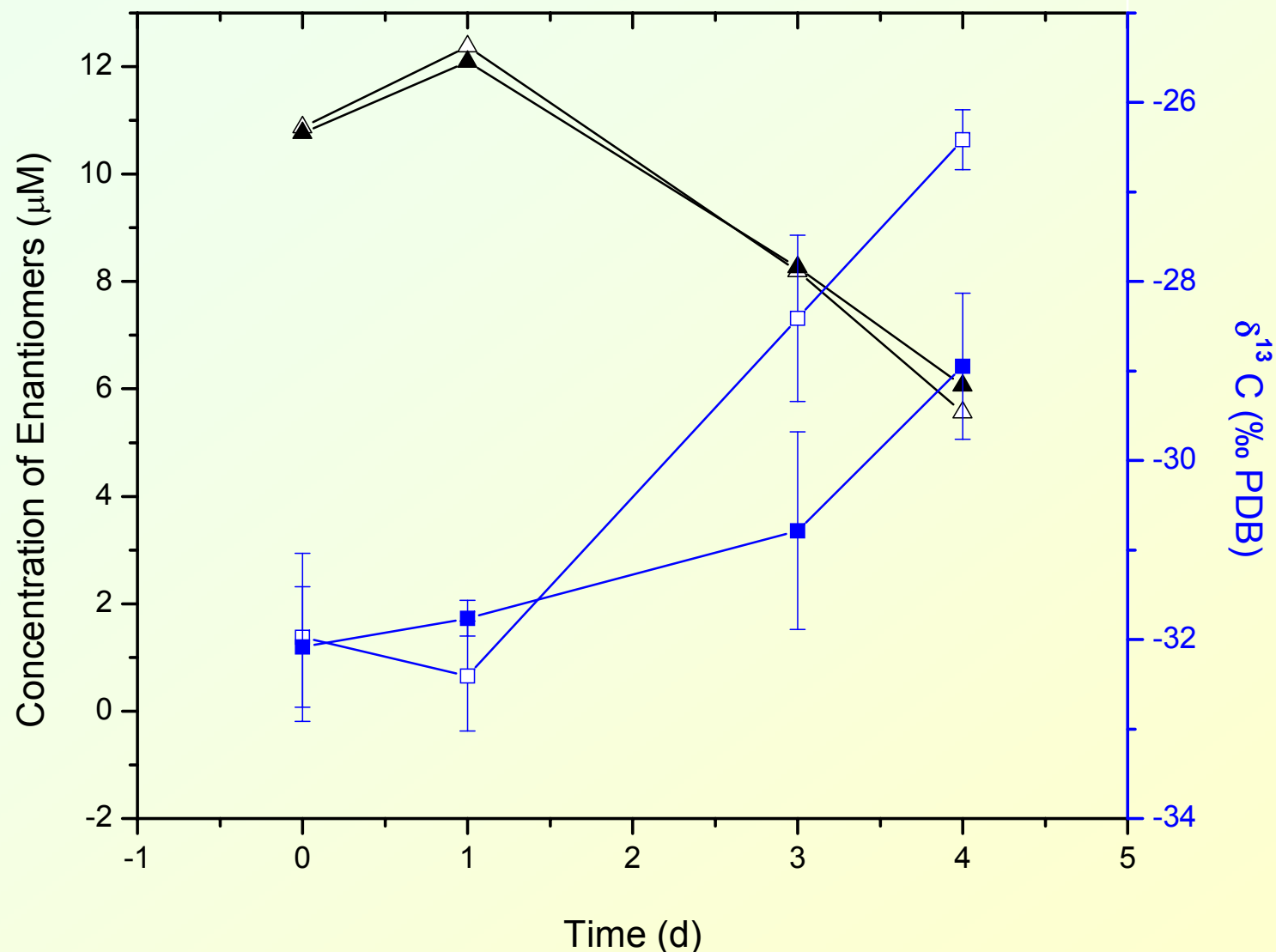




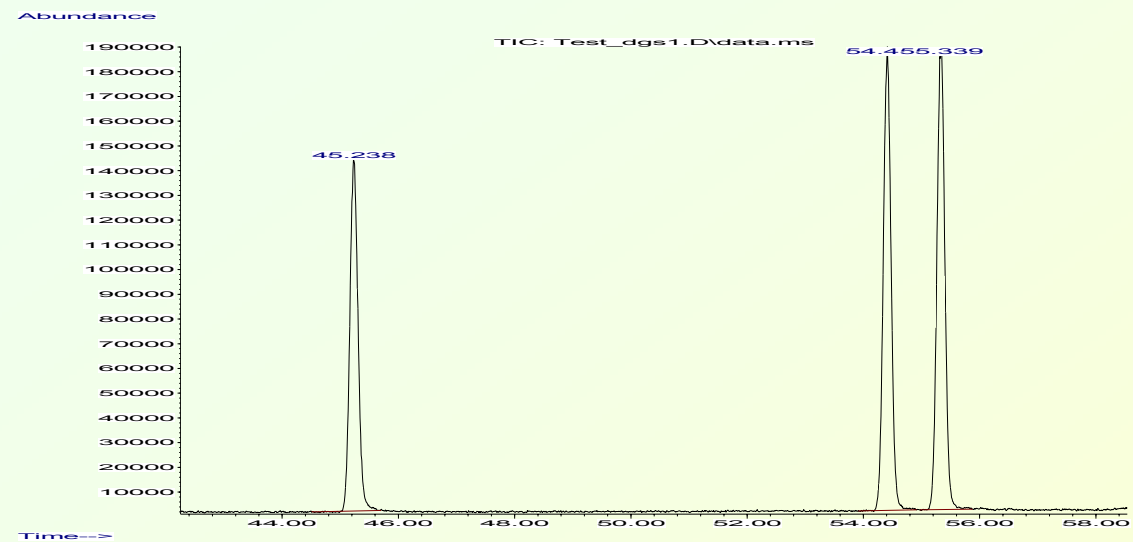
The Chiral GC-MS separation of alpha HCH and their metabolites

Alpha HCH's Biodegradation by *C. Pasteurianum*

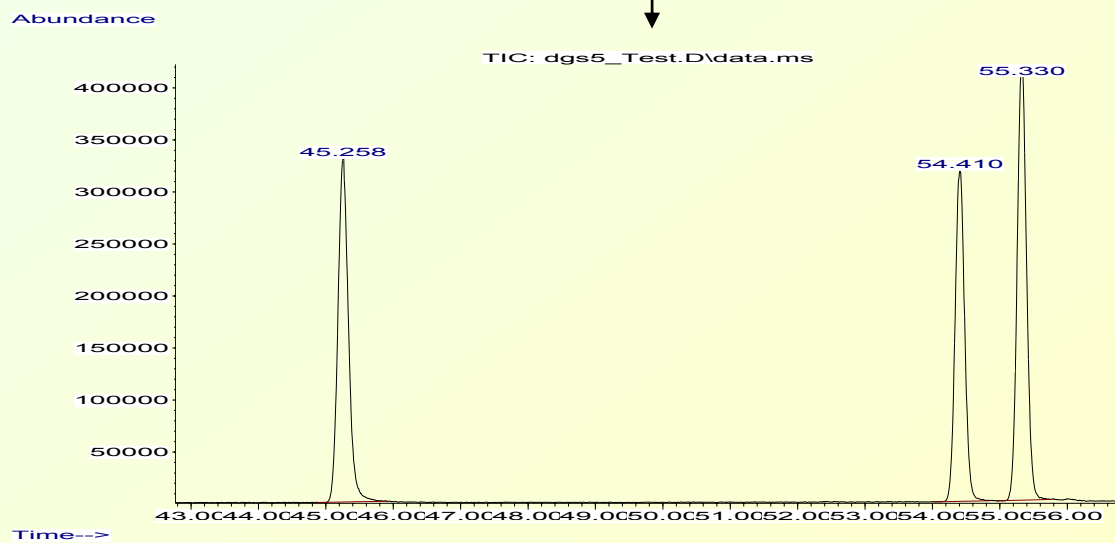




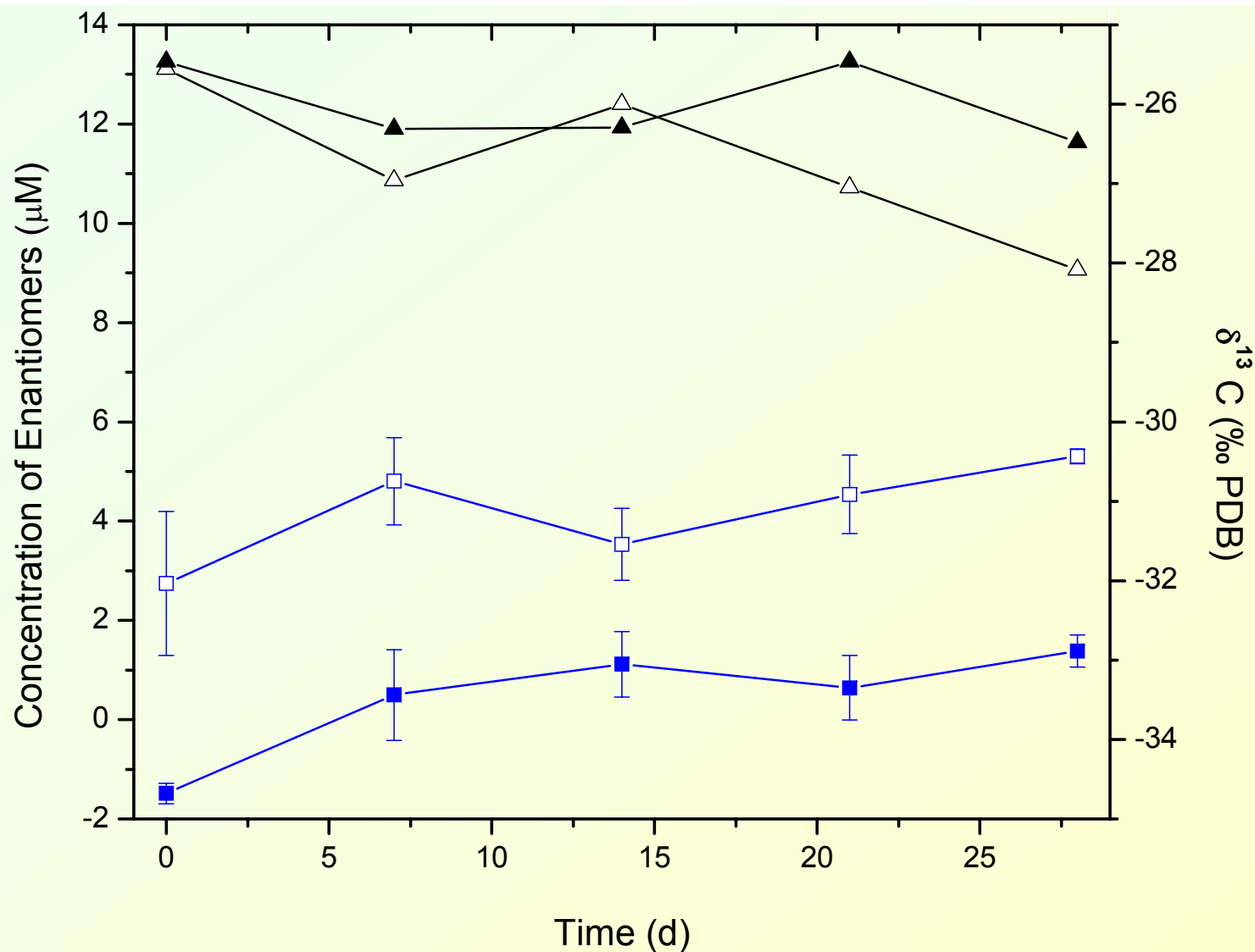
Change in concentration (Δ) and carbon isotope composition ($\square$) of (-) alpha HCH, in concentration ($\blacktriangle$) and carbon isotope composition ($\blacksquare$) of (+) alpha HCH during reductive dechlorination by *C. Pasteurianum*



28 Days of Incubation

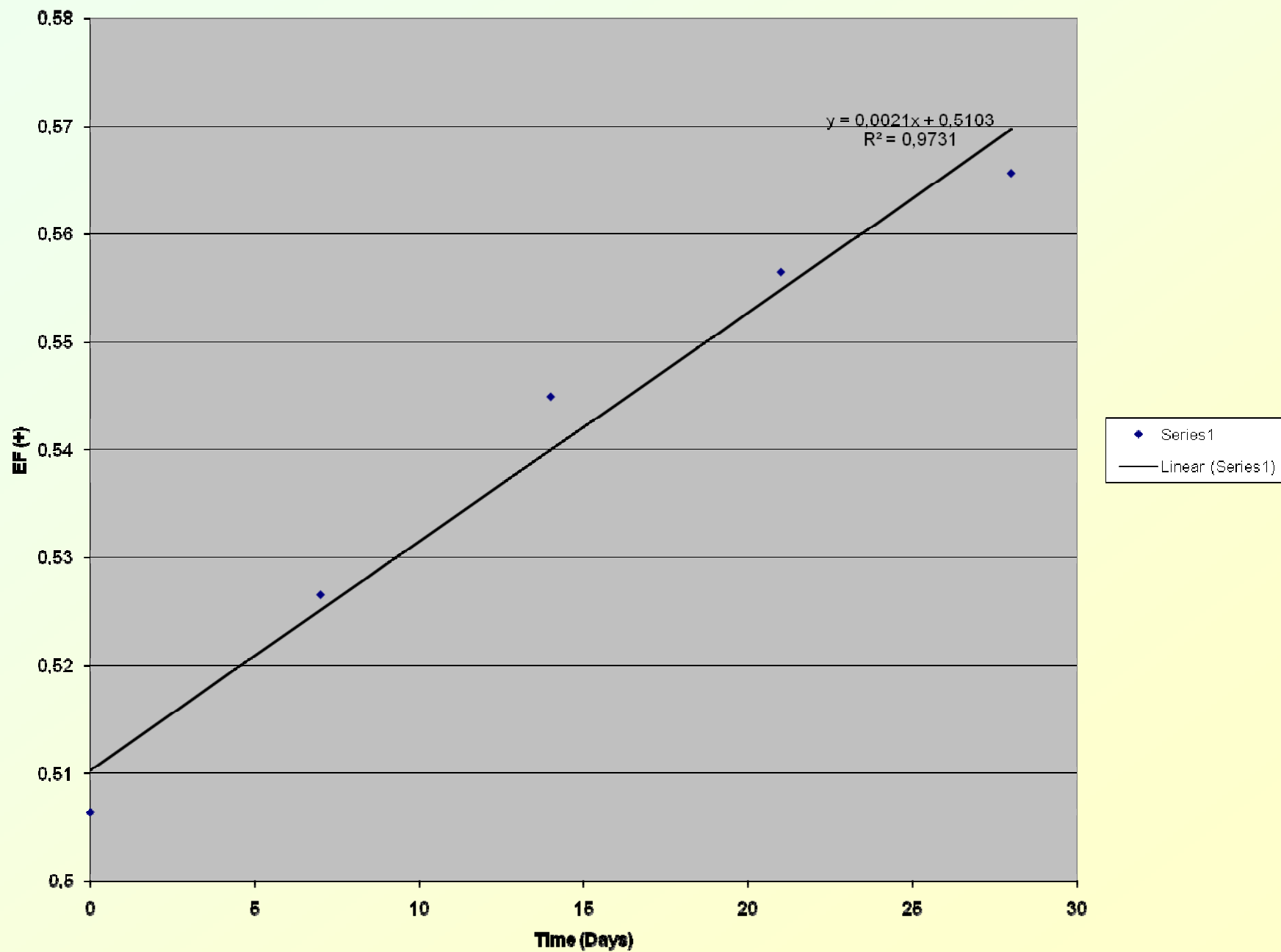


Enantioselective Transformation of alpha HCH by D. Gigas



Change in concentration (Δ) and carbon isotope composition ($\square$) of (-) alpha HCH, in concentration ($\blacktriangle$) and carbon isotope composition ($\blacksquare$) of (+) alpha HCH during reductive dechlorination by *D. Gigas*

Enantiomeric Fractionation of alpha HCH by D. Gigas



Conclusions

- The α HCH transformation by *Clostridium Pasteurianum* is associated with a low enantioselectivity with a specific isotope fractionation pattern for the enantiomers
- The α HCH transformation by *Desulfovibrio Gigas* is highly enantioselective even at a low extent of biotransformation
- Benzene, chlorobenzene and, the intermediate, δ -tetrachlorocyclohexene (chiral) were detected as microbial metabolites of alpha HCH transformation
- Combined Stable Isotope and Enantiomeric Fractionation of alpha HCH can be a new concept in Biogeochemistry of POPs
- More experimental data are needed to developed this approach

Acknowledgements

Environmental and Biogeochemistry Group,
Department of Chemistry,
Umeå University, Sweden



Department of Isotope Biogeochemistry,
Helmholtz Centre for Environmental Research
– UFZ Leipzig, Germany

