## Legacy and new POPs: What the North American Great Lakes can tell us.

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Indiana University

6th Network Conference on Persistent Organic Pollutant: 17–18 April, 2012



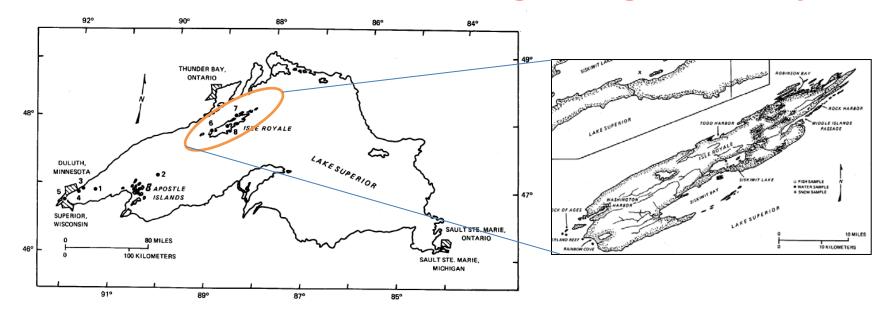
### Why the Great Lakes?

- Their total surface is  $\sim$ 250,000 km<sup>2</sup> (94,000 m<sup>2</sup>), and the total volume is 23,000 km<sup>3</sup> (5,500 mi<sup>3</sup>)
- The lakes are sometimes referred to as the North Coast or "Third Coast" of the US.

The Great Lakes hold 21% of the world's surface fresh

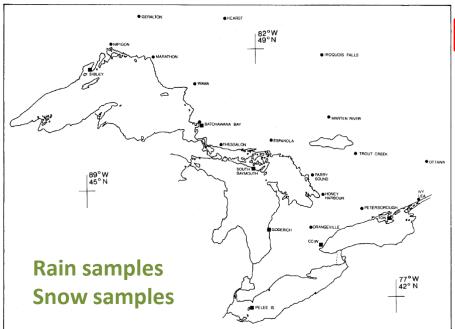


### It all started with the Great Lakes: first evidence of long range transport



PCBs and chlorinated pesticides were measured in fish, water and precipitation.

These findings were the first to suggest that atmospheric transport to Lake Superior was important.



#### Evidence was piling up

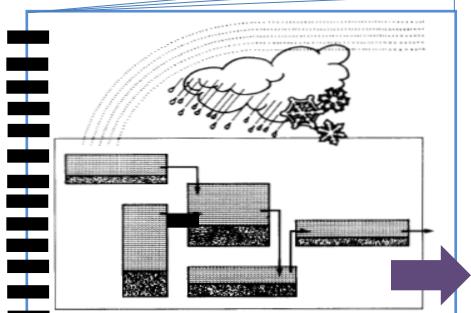
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#### CONCLUSIONS

PCBs and organochlorines are present in rain in the Great Lakes region in such levels as to give rise to great concern. It is apparent that atmospheric deposition is a major route for the introduction of the persistent substances among these materials into biological tissue, either through aquatic food chains or through deposition in the terrestrial comments of the latest through through the latest through th

Strachan & Huneault, J Great Lakes Res, 1979

### From science to policy: Annex 15 of the Great Lakes Water Quality Agreement



Summary Report of the Workshop on Great Lakes Atmospheric Deposition

Held October 29 - 31, 1986

The Guild Inn, Scarborough, Ontario

Sponsored by the International Joint Commission's Great Lakes Science Advisory Board Water Quality Board, and International Air Quality Advisory Board



ANNEX 15 - AIRBORNE TOXIC
SUBSTANCES

Called for the Integrated
Atmospheric Deposition
Network

#### Why Great Lakes air now?

- Atmospheric transport is now <u>the</u> primary transport pathway for the input of toxic and persistent organic pollutants (POPs) into the Great Lakes.
- Air concentrations respond rapidly to changes in emissions, but it is not known how rapidly the lakes and the fishes in them will respond
- Increasing number of fish consumption advisories



### Integrated Atmospheric Deposition Network (IADN)

- Joint EPA-Environment Canada project
  - Grantee is Prof. Hites at Indiana University
- Measure pollutants in air and precipitation around the Great Lakes (U.S. runs five sites)
- Goals:
  - Determine atmospheric loadings to the lakes
  - Look at trends in levels (spatial and temporal)
- Strengths:
  - Long term (running continuously since 1990)
  - Good spatial coverage
  - Legacy and emerging compounds

### Comparison of long term air programs



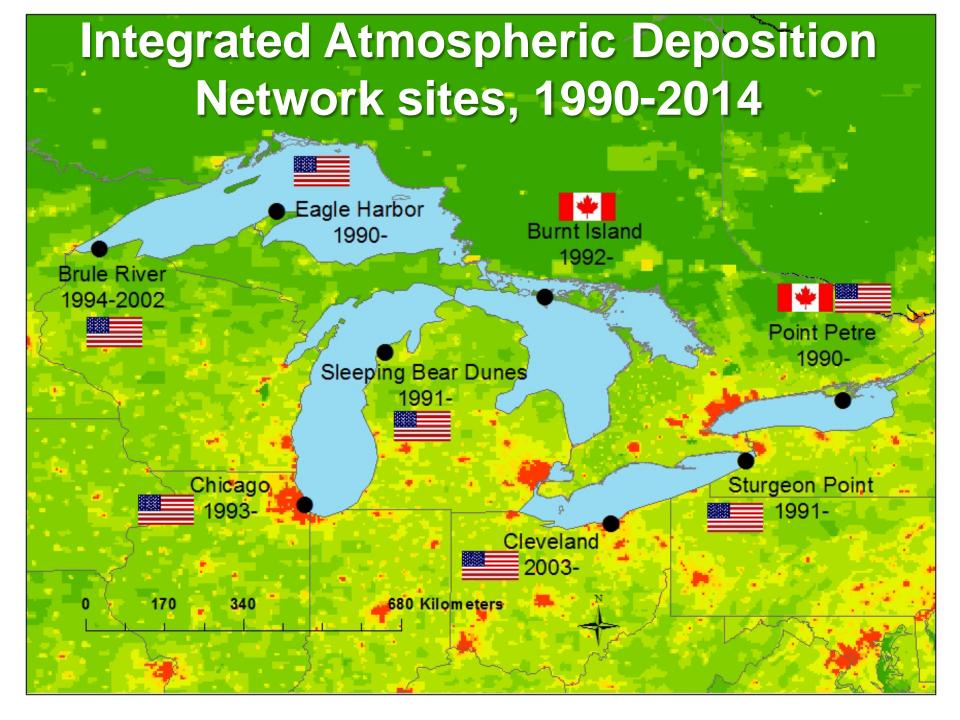




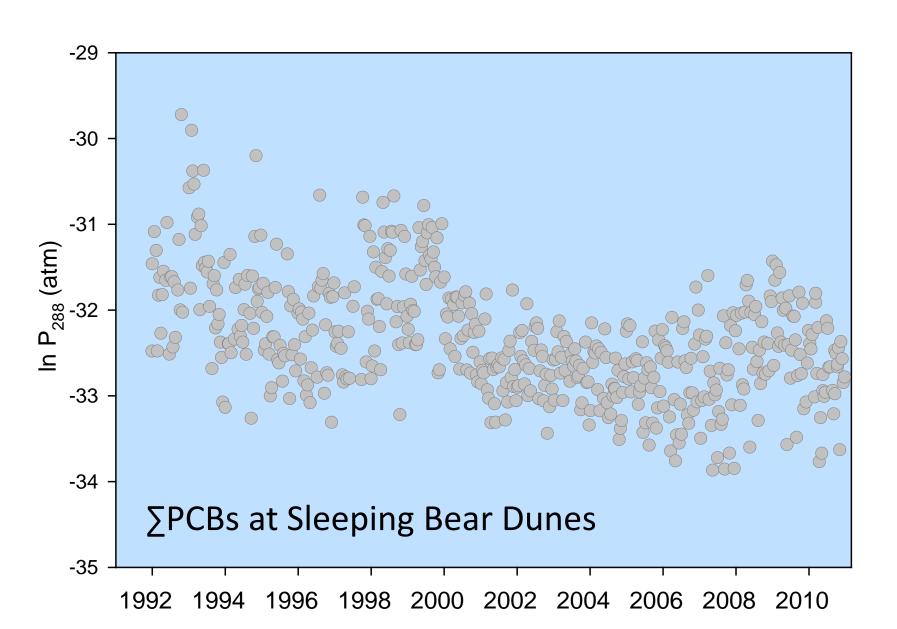
	TOMPS	EMEP	IADN	AMAP
Location	UK	EU (NILU)	USA	Arctic
Starting date	1990	1991*	1990	1992 <del>+</del>
N. sites	6	22*	6	4+8
Analytes	PCDD/Fs, PCBs, PAHs	PCBs, PAHs, OCPs, PBDEs	PCBs, PAHs, OCPs, PBDEs	PCBs, OCPs, PBDEs

<sup>\* 1991: 1; 1995:5; 2000:8; 2005:11; 2009:22</sup> 

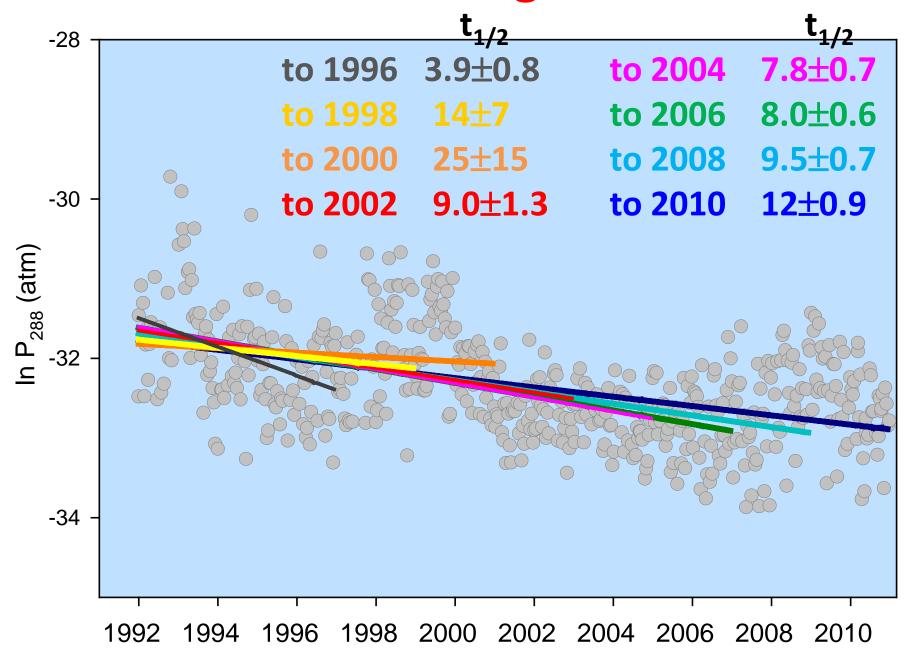
<sup>+ 1992:</sup> Alert; 1993: Zeppelin



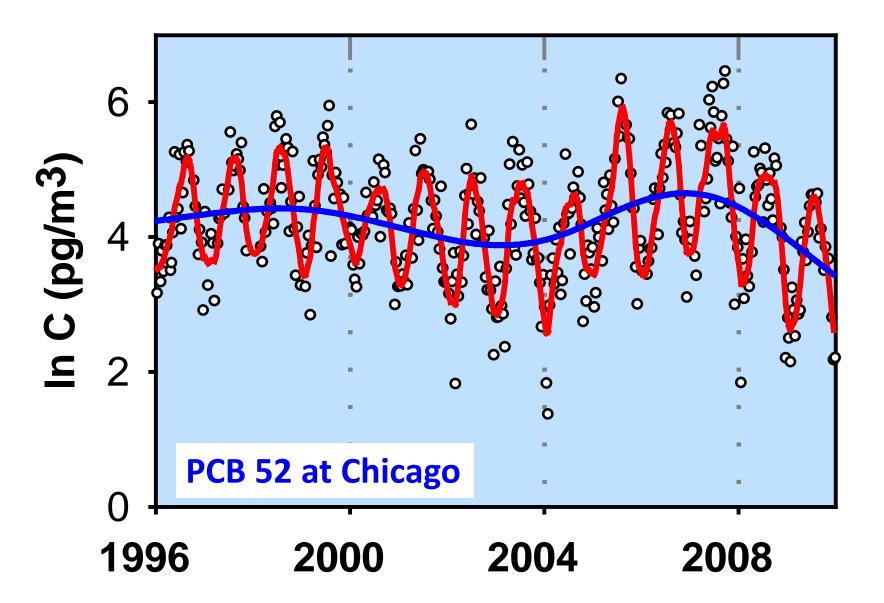
#### The need for long term data

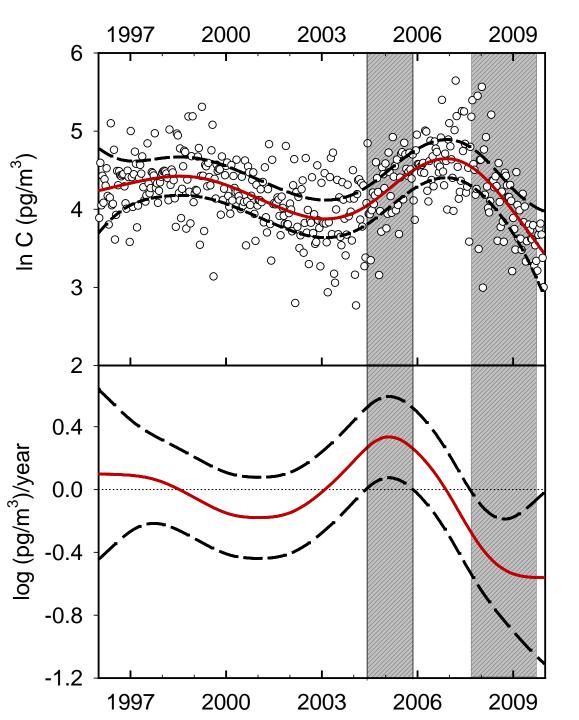


#### The need for long term data



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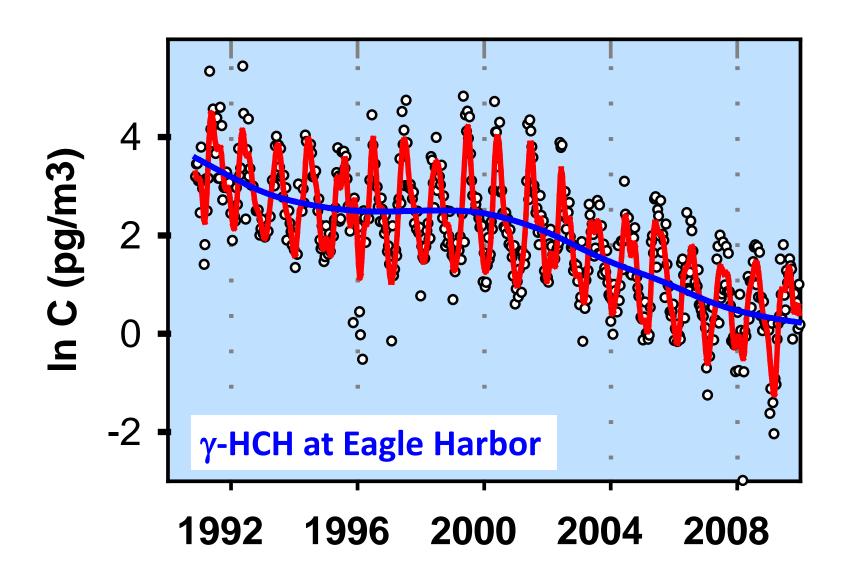




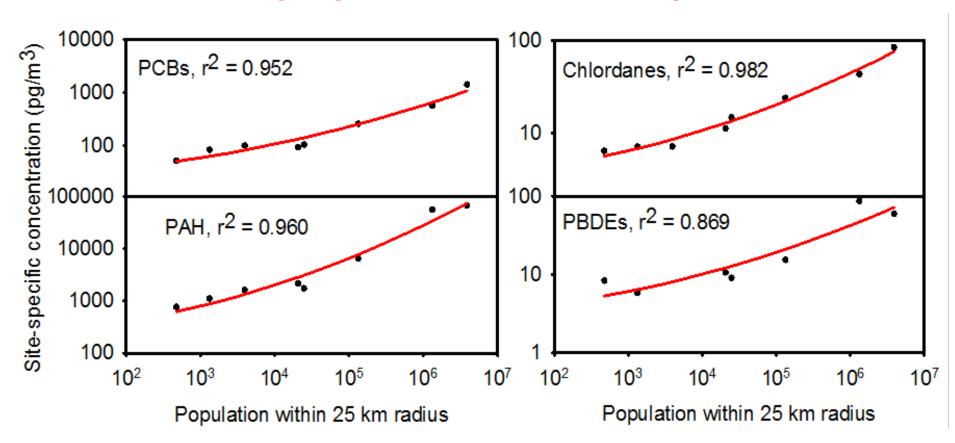
# PCB 52 at Chicago: what happened around 2005?



### The atmosphere shows rapid declines for banned chemicals



### Atmospheric concentrations track population density



Other  $r^2$  values: DDTs=0.774; Endosulfans=0.314;  $\alpha$ -HCH=0.128; lindane=0.098

### What's Going On?

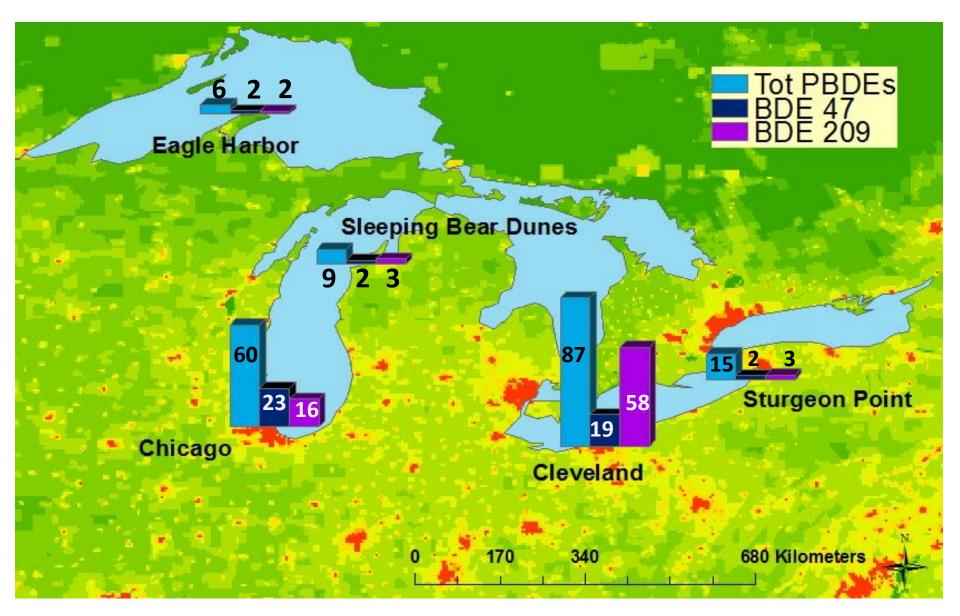
	Half- life	Primary Control	Secondary Control	Date Banned
•	(years)			
$\Sigma$ PCB	20+	Urbanization	Seasonality	1977
$\Sigma$ PAH	20	Urbanization		In use
$\alpha$ -HCH	4	Time		1980
$\gamma$ -HCH	5	Seasonality	Time	2004/2009
$\Sigma$ DDT	8	Urbanization	Seasonality	1972
$\Sigma$ Endosulfans	10	Seasonality		2016
$\Sigma$ Chlordanes	13	Urbanization	Seasonality	1980

### What about emerging compounds?

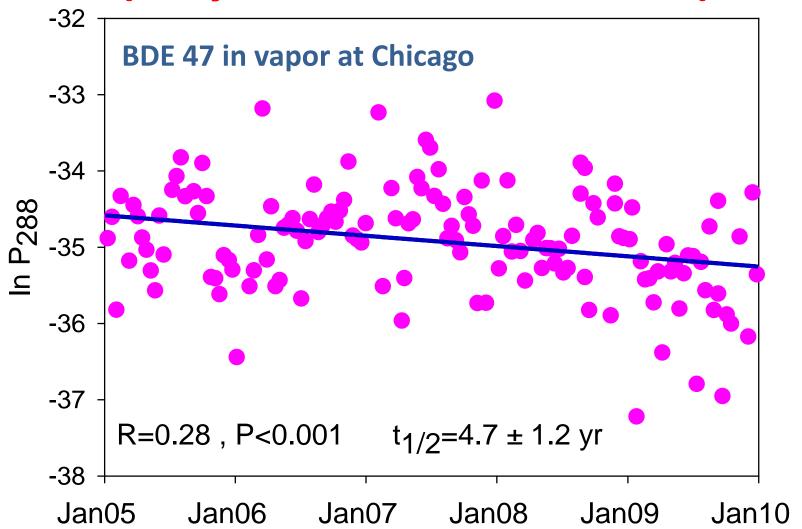
- Starting from 2004, PBDEs were added to the list of target compounds.
- They were officially included in IADN in 2009.
- The list of brominated flame retardants (BFR) is ever expanding.



### Levels of PBDEs in air (pg/m³)



### PBDEs levels in the air are going down (only the abandoned ones)



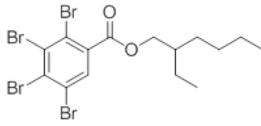


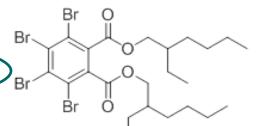
It's like a
whack-a-mole
game
(with chemicals)

### Replacements for penta BDEs: Firemaster 550

#### Components:

- 1. Triphenyl Phosphate
- 2. Triaryl **phosphate** isopropylated
- 3. Bis(2-ethylhexyl) tetrabromophthalate
- 4. 2-ethyl hexyl 2,3, 4, 5-tetra**bromo**benzoate







#### Technical Information

www.Chemtura.com Effective: 11.01.2007

#### Firemaster® 550

Phosphorus-Bromine Flame Retardant

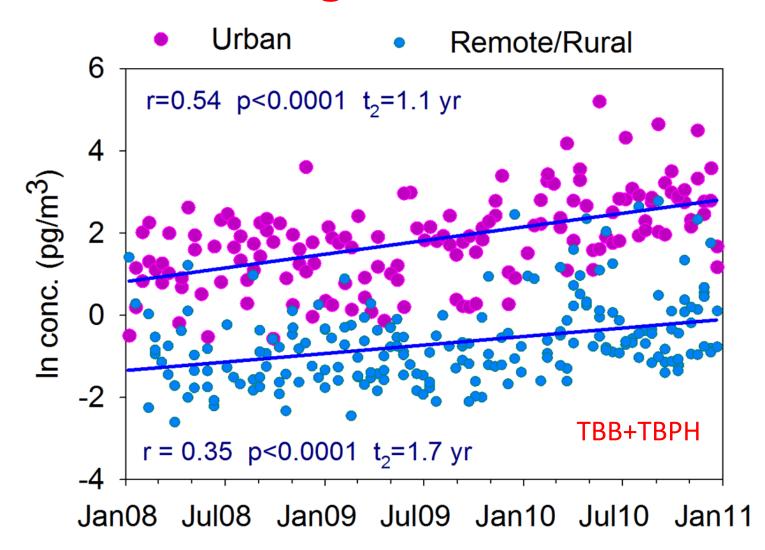
**Firemaster 550** is a low viscosity liquid flame retardant for polyurethane foams and other applications, which require a liquid product. Firemaster 550's high efficiency as a flame retardant is a result of phosphorus-bromine synergy. It is based on proprietary Chemtura technology and does not contain brominated diphenyl ethers.

Phosphorus-Bromine Flame Retardant CAS Reg. Number Proprietary

### Levels of tetrabromo esters in air are close to those of PBDEs



### Levels of tetrabromo esters are increasing in the Great Lakes' air



#### **Alternative FR: Decabromodiphenylethane (DBDPE)**

- DBDPE has been marketed as general purpose substitute for BDE-209. Produced since the 1990s as SAYTEX 8010.
- It is an additive flame retardant with similar applications to those of BDE209: acrylonitrile-butadiene-styrene (ABS), high impact polystyrene (HIPS) plastics, and textile backcoating)
- Not much data on toxicology, especially from independent sources (several published papers are from sources related to Albemarle)

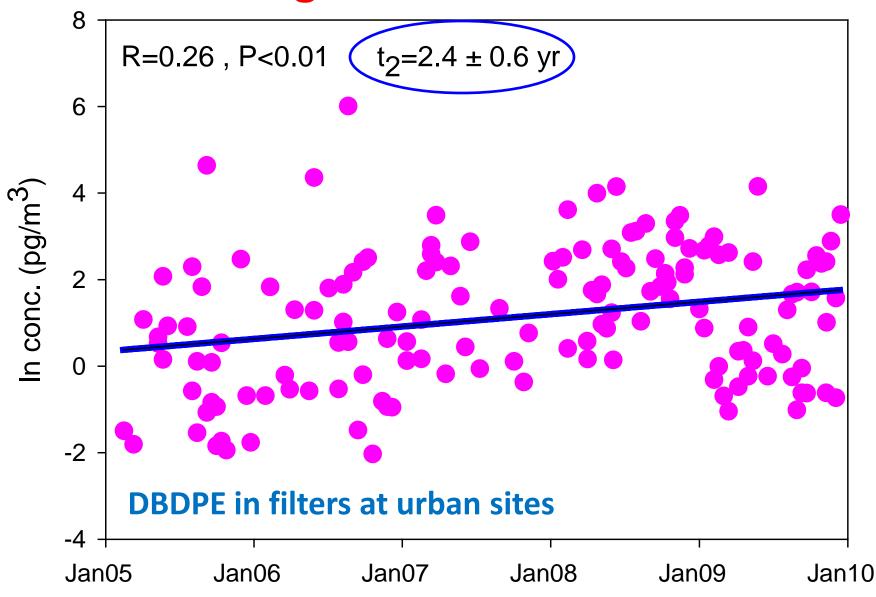
#### **Decabromodiphenylethane (DBDPE)**



Ubiquitous and worldwide pollutant!

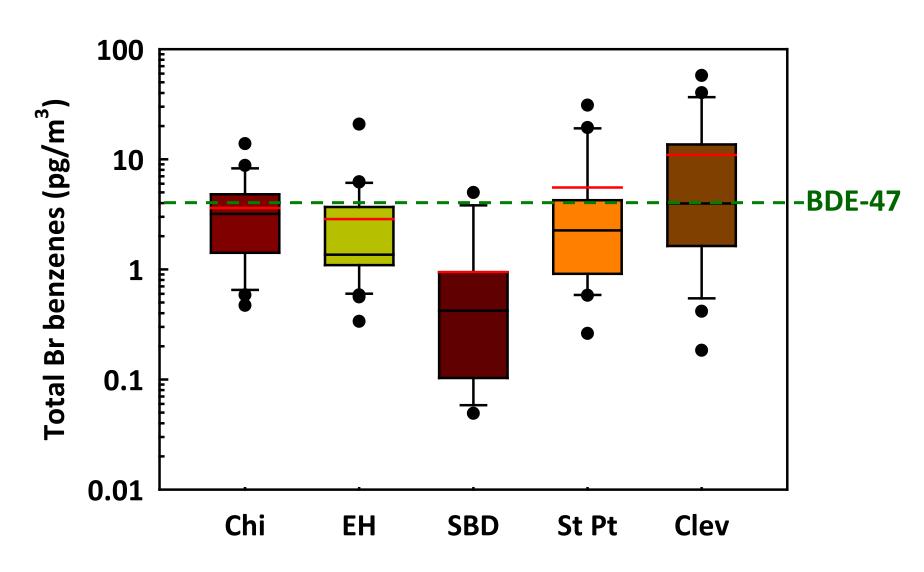
Rickund, 2008; Hu, 2008; Gauthier, 2009; Harrad, 2008

### Decabromodiphenylethane (DBDPE) levels increasing in the Great Lakes' air

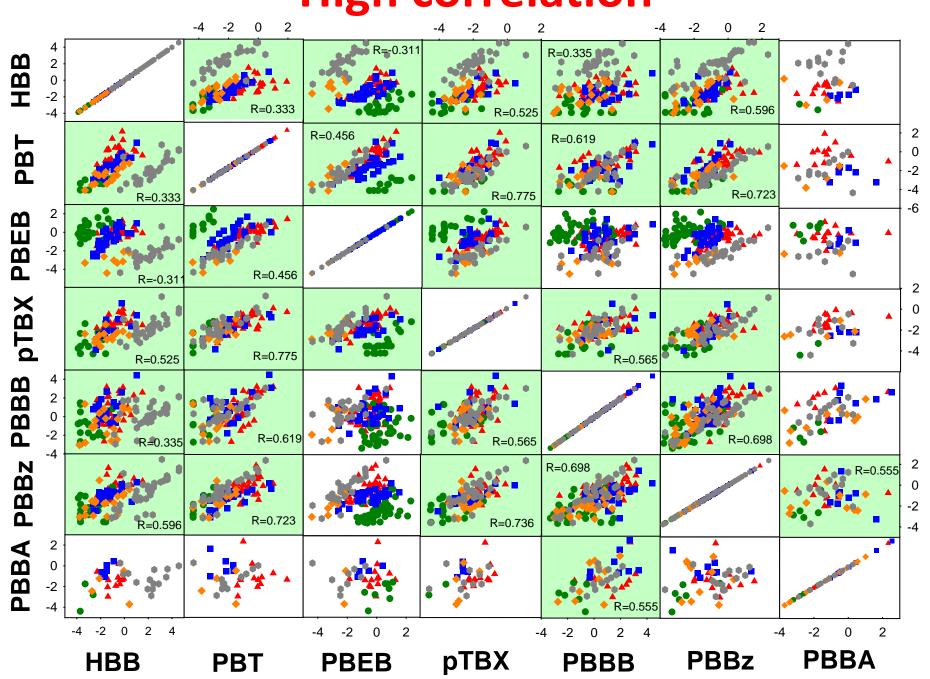


### Bromobenzenes: a puzzling group of compounds

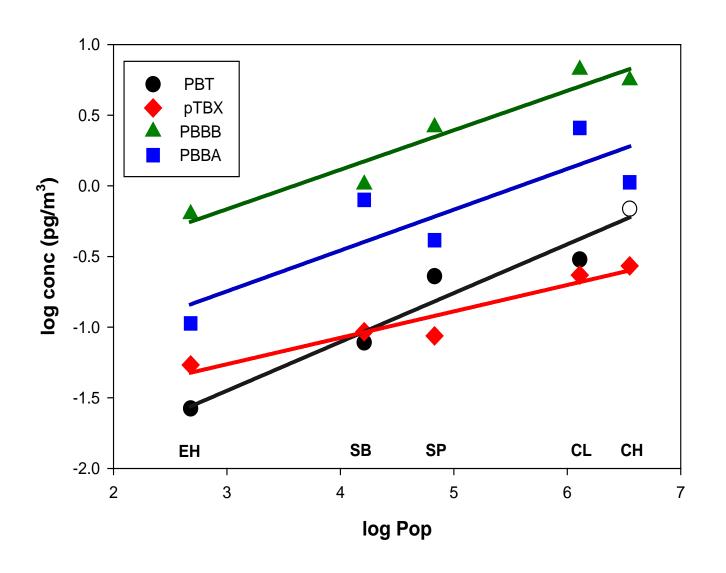
### Total concentrations of the 6 brominated benzenes in Great Lakes air (vapor phase)



**High correlation** 

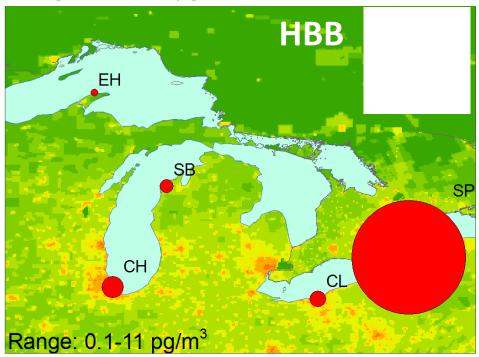


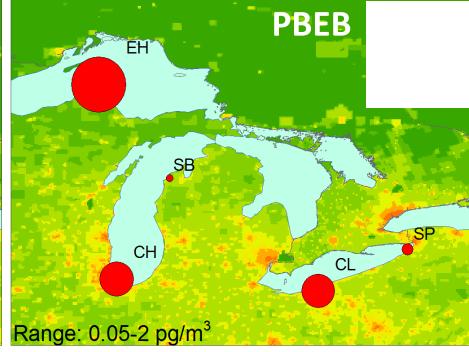
### Relationship with population

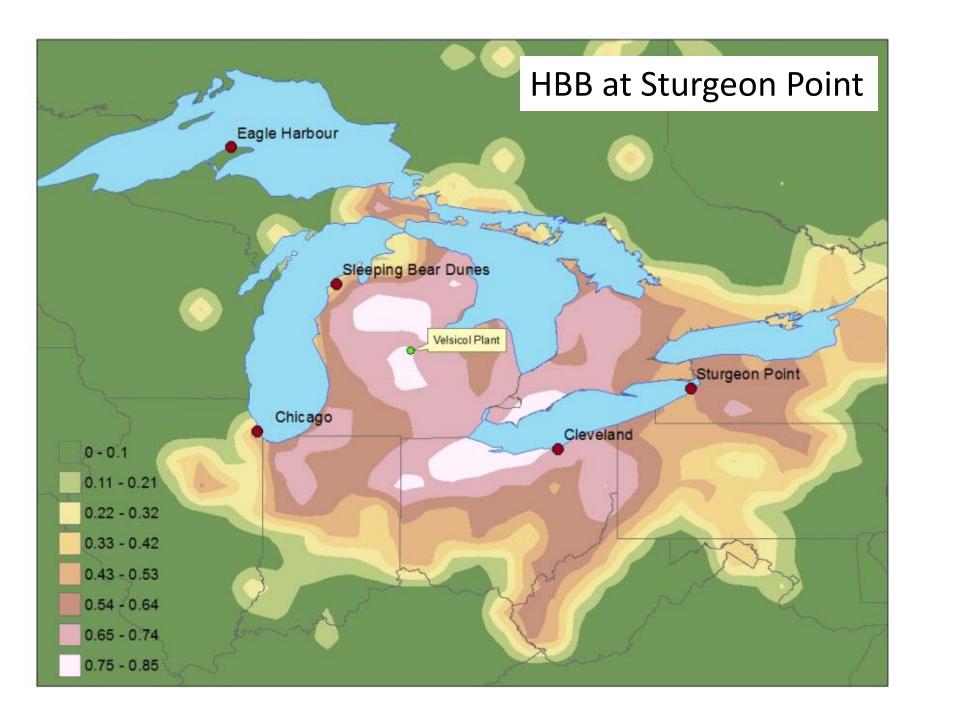


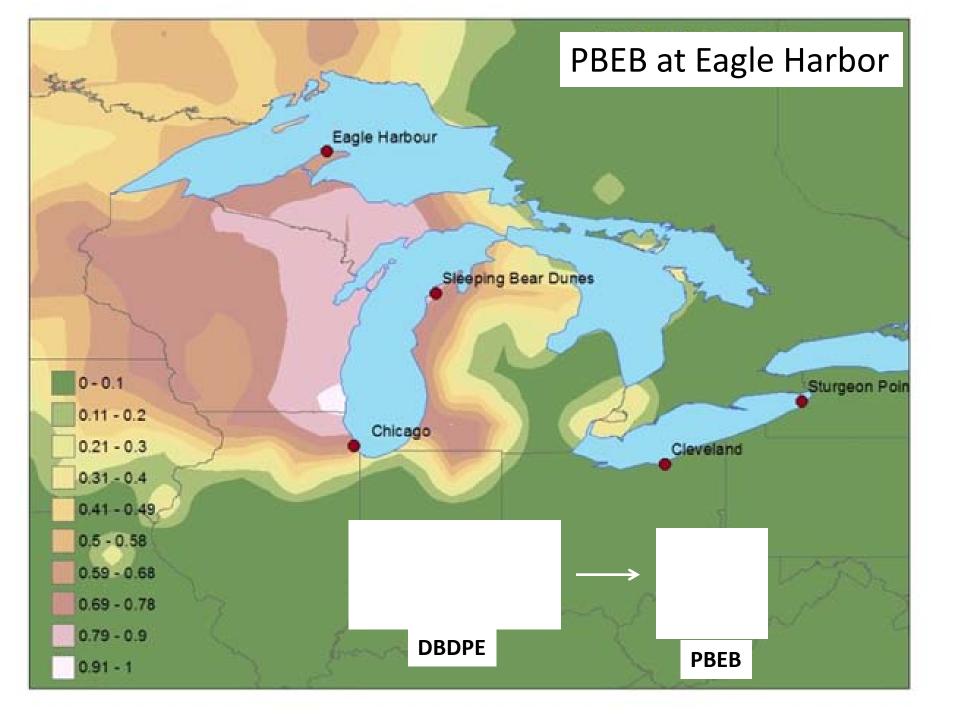
### **PBT** CH CL Range: 0.03-0.3 pg/m<sup>3</sup>

### High urban effect vs Localized sources









#### **Conclusions**

- We should use the lessons learned from legacy compounds to predict the behavior of emerging compounds, especially for temporal trends.
- Air responds quickly to changes in emissions
- Urban locations are generally sources of POPs
- We need good spatial coverage to obtain an exhaustive picture

### Acknowledgements

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